

Abstract Submitted  
for the 4CF17 Meeting of  
The American Physical Society

**Octahedral Tilting from Organic Cations in Defect-Ordered Hybrid Perovskites**<sup>1</sup> ANNALISE MAUGHAN, Colorado State Univ, ALEX GANOSE, University College London, ANDREW CANDIA, JULIETTE GRANGER, Colorado State Univ, DAVID SCANLON, University College London, JAMES NEILSON, Colorado State Univ — Perovskite semiconductors such as methylammonium lead iodide ( $\text{CH}_3\text{NH}_3\text{PbI}_3$ ) are a technologically-relevant family of materials, finding applications in field-effect transistors, light-emitting diodes, and low-cost photovoltaics, yet materials such as methylammonium lead iodide present significant concerns for toxicity and material stability. Defect-ordered perovskites provide a platform to study optical and electronic behavior in materials with improved toxicity and stability outlooks relative to lead-containing counterparts. We have prepared the series of vacancy-ordered double perovskites  $A_2\text{SnI}_6$ , where  $A = \text{Cs}^+$ ,  $\text{CH}_3\text{NH}_3^+$ , and  $\text{CH}(\text{NH}_2)_2^+$ . While  $\text{Cs}_2\text{SnI}_6$  exhibits moderate electronic conductivity, incorporation of  $\text{CH}_3\text{NH}_3^+$  and  $\text{CH}(\text{NH}_2)_2^+$  reduces carrier mobility across the series. We correlate the observed trends in carrier mobility with rotational disorder and anharmonicity of the  $[\text{SnI}_6]$  octahedral units, which is enhanced through hydrogen bond formation in the hybrid  $(\text{CH}_3\text{NH}_3)_2\text{SnI}_6$  and  $(\text{CH}(\text{NH}_2)_2)_2\text{SnI}_6$  compounds. These studies provide an avenue to explore the influence of organic-inorganic coupling on the properties of perovskite-based materials for emerging optical and electronic applications.

<sup>1</sup>This research is supported by the grant DE-SC0016083 funded by the U.S. Department of Energy, Office of Science.

Annalise Maughan  
Colorado State Univ

Date submitted: 19 Sep 2017

Electronic form version 1.4