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Gibbs effect on evaporation of a vapor bubble.¹ ARASH ASADOL-LAHI, ASGHAR ESMAEELI, Southern Illinois University Carbondale — In numerical simulations of boiling, the interface temperature is generally assumed to be the same as the saturation temperature at the system pressure. While this is a reasonable assumption for stable boiling, it may not be correct for rapid evaporation, where the highly unsteady nature of the physics can lead to the deviation of the interface temperature from the saturation temperature. In this study we explore the effect of the interface temperature on the stable and unstable growth of a vapor bubble in a superheated liquid. To this end, we use a comprehensive Gibbs-Thomson equation to determine the interface temperature and explore the effect of local variation of this parameter on the bubble dynamics.

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Arash Asadollahi Southern Illinois University Carbondale

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