## Abstract Submitted for the GEC13 Meeting of The American Physical Society

A spectroscopic study of ethylene destruction and by-product generation using a three-stage atmospheric packed-bed plasma reactor MARKO HUEBNER<sup>1</sup>, OLIVIER GUAITELLA<sup>2</sup>, ANTOINE ROUSSEAU<sup>3</sup>, JUER-GEN ROEPCKE<sup>4</sup>, None — Using a three-stage dielectric packed-bed plasma reactor at p = 1 bar the destruction of  $C_2H_4$  and the generation of major by-products have been studied by FTIR spectroscopy. As test gas mixture air containing 0.12% humidity with 0.1% ethylene admixture was used. In addition to the fragmentation of the precursor gas, the evolution of the concentration of ten stable reaction products, CO, CO<sub>2</sub> O<sub>3</sub>, NO<sub>2</sub>, N<sub>2</sub>O, HCN, H<sub>2</sub>O, HNO<sub>3</sub>, CH<sub>2</sub>O and CH<sub>2</sub>O<sub>2</sub> has been monitored. Applying three sequentially working discharge cells (f = 4 kHz, U = 9 - 12 kV) a nearly complete decomposition of C<sub>2</sub>H<sub>4</sub> could be achieved. In maximum the specific energy deposition was about 900 Jl<sup>-1</sup>. The value of the specific energy  $\beta$ , characterizing the energy efficiency of the ethylene destruction in the used reactor, was 330  $Jl^{-1}$ . The carbon balance of the plasma chemical conversion of ethylene has been analyzed. As a main result of the study, the application of three reactor stages suppresses essentially the production of harmful by-products as formaldehyde, formic acid and  $NO_2$  compared to the use of only one or two stages.

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