

Abstract Submitted
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Geometry and Electronic Structure of Nickel Benzene Complexes

A. K. KANDALAM, S. N. KHANNA, PURU JENA, Virginia Commonwealth University — A synergistic approach combining the negative ion photo-detachment spectroscopy and the first principles density functional calculations is used to probe the geometrical structure, nature of electronic bonding and the magnetic moment of anionic and neutral nickel-benzene (Bez) complexes. It is shown that, contrary to the previously published results, the ground state of a Ni-Bez anion is a structure where the Ni atom is inserted in one of the C-H bonds of the benzene molecule. However, a detailed investigation of the reaction barrier shows that this ground state may be inaccessible in ordinary experiments and the observed structures in negative ion photo-detachment spectra may be the higher energy structure where a Ni atom occupies a hollow site above the benzene molecule. The stability of inserted structures in multiple Ni-Bez complexes and how they may be attained in carefully planned experiments will be highlighted.

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