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**Theoretical Investigation on the Electronic Structure of Alq<sub>3</sub>/Al Interface** SUSUMU YANAGISAWA, Osaka University, YOSHITADA MORIKAWA, Osaka Univ., RICS-AIST — Alq<sub>3</sub> [tris-(8-hydroxyquinolato) aluminum] is one of the most widely used electron transport and emissive material in organic light-emitting devices (OLEDs). From the experimental observation of an extra gap state at the Alq<sub>3</sub>/Al interface, a strong chemical interaction between the Alq<sub>3</sub> molecule and the Al surface was suggested. Contrary to the experimental studies, previous DFT calculations concluded that the interaction was physisorptive. One possible reason for the discrepancy between the theoretical and the experimental results is the complexity of the experimentally used electrode surfaces. In the present study, we investigated the effect of the surface roughness on the electronic properties of the Alq<sub>3</sub>/Al interface by examining various possible electrode structures. We examined three structures for the Al substrate, the flat Al(111) surface, the Al(332) stepped surface, and the Al adatom adsorbed Al(111) surface. Alq<sub>3</sub> molecules are bound to Al substrates through their O atoms and about 0.3-0.6 electrons are transferred from the Al substrates to Alq<sub>3</sub>. Upward configurations, in which molecular permanent dipole moments are directed to the vacuum side, reduce the work function by 1.0-1.5 eV, in reasonable agreement with experimental results. The characteristic of the molecular orbitals of Alq<sub>3</sub> were kept upon adsorption, which seems inconsistent with the gap state derived from the interfacial chemical interaction observed in the UPS and MAES experiments. Further details will be presented.

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