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**Crystallization Behavior of Chemically Prepared Nanoparticles of Amorphous Fe-B\*** ERIK PEARSON, ADRIAN GOREA, THOMAS EKIERT, KARL UNRUH, University of Delaware — The crystallization behavior of amorphous Fe-B nanoparticles prepared by reducing an aqueous solution of  $\text{Fe}^{+2}$  ions with  $\text{NaBH}_4$  has been studied by differential scanning calorimetry (DSC), vibrating sample magnetometry (VSM), and x-ray diffraction (XRD) measurements. At a heating rate of  $10^\circ\text{C}/\text{min}$  the DSC measurements show a sharp and well defined exothermic peak at a temperature of about  $475^\circ\text{C}$  and a Kissinger analysis of the shift in the position of this peak as a function of the heating rate yields an activation energy of about  $3.6\text{ eV/at}$ . The VSM measurements also exhibit a sharp increase in the magnetization at a temperature  $475^\circ\text{C}$  (at a heating rate of  $10^\circ\text{C}/\text{min}$ ). X-ray diffraction measurements on samples heated to temperatures slightly above  $475^\circ\text{C}$  verify that the observed DSC and VSM signals correspond to the transformation from the as-prepared amorphous structure to a crystalline structure. \* This work has been supported by AFRL DARPA METAMATERIALS contract no. F33615-01-2-2166, ARO DEPCOR grant no. W911NF-04-1-0264, and the Undergraduate Research Program at the University of Delaware.

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