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**Light Scattering Study of a fluorinated Alkyl Methacrylate Polymer in Carbon Dioxide** JI GUO, APS, ACS, JOSEPH M. DESIMONE, MICHAEL RUBINSTEIN — The solution properties of fluorinated homopolymers in liquid and supercritical carbon dioxide were studied by light scattering. Poly(fluoroalkyl methacrylate) samples were fractionated by carbon dioxide (CO<sub>2</sub>) to achieve narrow polydispersity and the refractive index of each sample in CO<sub>2</sub> was measured with a high pressure optical setup. Molecular weight, size, and interaction parameters of the polymer dissolved in CO<sub>2</sub> were studied as a function of temperature and CO<sub>2</sub> density using both static and dynamic light scattering. The solvent quality of CO<sub>2</sub> was shown to quantitatively improve with temperature and CO<sub>2</sub> density. We observed both  $\theta$ -temperature and  $\theta$ -density for poly(fluoroalkyl methacrylate) in CO<sub>2</sub>. The hydrodynamic radius was found to increase with the temperature and density of CO<sub>2</sub>. We demonstrate that the second virial coefficient of the polymer in CO<sub>2</sub> can be expressed in terms of the universal interaction parameter in the good solvent regime. This confirms that polymers in CO<sub>2</sub> have the same universal behavior as in organic solvents.

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