

Abstract Submitted
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Molecular dynamics study of the Mg - AMP interaction in water DON BRUGESS, Asbury College, IOAN KOSZTIN, University of Missouri - Columbia — The conformational states of Mg^{++} complexed with adenosine monophosphate (AMP) in water solution is analyzed using classical molecular dynamics (MD) simulations. Relevant portions of the 3D *potential of mean force* (PMF) of Mg^{++} is reconstructed by employing two distinct methods: (1) equilibrium MD simulations using the umbrella sampling and weighted histogram analysis method, and (2) non-equilibrium steered MD simulations using a recently developed method based on the Jarzynski equality. Withing computational errors, both methods yield the same results. Two topologically distinct sets of equilibrium conformations of Mg^{++} in the vicinity of the phosphate moiety are found. The free energy difference between the states within a given set is about $1\ kT$, and are separated by potential barriers of $\sim 10\ kT$ in height. However, the free energy difference between states from the two distinct sets are found to be unrealistically high ($\sim 10\ \text{kcal/mol}$). Also, the calculated dissociation energy of Mg^{++} from AMP exceeds several times the corresponding experimental value. Possible sources of this discrepancy are discussed, and alternative methods to improve the accuracy of the calculations are proposed.

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