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**Rings, Towers and Cages in  $Zn_nO_n$  Clusters** ARTHUR REBER, SHIV KHANNA, Department of Physics, Virginia Commonwealth University, Richmond Va. 23284, JAGTAR HUNJAN, MARCELA BELTRAN, Instituto de Investigaciones en Materiales, Universidad Nacional Autonoma de Mexico — It is shown that the transition from an elementary ZnO molecule to compact bulk wurtzite ZnO proceeds via hollow rings, towers, and cages. First principles electronic structure calculations within a gradient corrected density functional framework have been carried out to investigate the progression of geometries and electronic properties of  $Zn_nO_n$  ( $n=2-12,15,16,21$ ) clusters. It is shown that  $Zn_nO_n$  ( $n=2, 3, 4, 5, 6, 7$ ) clusters are all single, highly stable rings and that  $Zn_3O_3$  is particularly stable. Starting at  $Zn_8O_8$ , these elementary rings begin to assemble into column structures that begin to distort at  $n=10$ . The ground states of  $Zn_{12}O_{12}$ , and  $Zn_{16}O_{16}$  are single cages while the structure of  $Zn_{11}O_{11}$ ,  $Zn_{15}O_{15}$  and  $Zn_{21}O_{21}$  can be described as barrels. The  $Zn_{12}O_{12}$  cage has a high dissociation energy and a large highest occupied molecular orbital (HOMO)-lowest unoccupied molecular orbital (LUMO) gap of 2.51 eV making it a potential candidate for cluster assemblies.

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