Abstract Submitted for the MAR07 Meeting of The American Physical Society

Cohesion of BaReH<sub>9</sub> and BaMnH<sub>9</sub>: Density Functional Calculations and Prediction of  $(MnH_9)^{2-}$  Salts<sup>1</sup> DAVID SINGH, Oak Ridge National Laboratory, M. GUPTA, Universite Paris-Sud, R. GUPTA, CEA, Saclay — Density functional calculations are used to calculate the structural and electronic properties of BaReH<sub>9</sub> and to analyze the bonding in this compound. This compound has an exceptionally high H to metal ratio of 4.5. The high coordination of Re in BaReH<sub>9</sub> is due to bonding between Re 5d states and states of d-like symmetry formed from combinations of H s orbitals in the H<sub>9</sub> cage. This explains the structure of the material, its short bond lengths and other physical properties, such as the high band gap. We compare with results for hypothetical BaMnH<sub>9</sub>, which we find to have similar bonding and cohesion to the Re compound. This suggests that it may be possible to synthesize  $(MnH_9)^{2-}$  salts. Depending on the particular cation, such salts may have exceptionally high hydrogen contents, in excess of 10 weight %.

<sup>1</sup>Work at ORNL is supported by DOE, DMS&E.

David Singh Oak Ridge National Laboratory

Date submitted: 30 Oct 2006

Electronic form version 1.4