## Abstract Submitted for the MAR07 Meeting of The American Physical Society

Exchange constants and second-order magnetic anisotropy in cyanide-bridged Fe2M2 single-molecule magnets<sup>1</sup> KYUNGWHA PARK, Virginia Tech, STEPHEN HOLMES, University of Kentucky — Electronic structure and intramolecular exchange constants are calculated for three cyanide-bridged single-molecule magnets, [Tp\*Fe<sup>III</sup>(CN)<sub>3</sub>M<sup>II</sup>(DMF)<sub>4</sub>]<sub>2</sub>(OTf)<sub>2</sub>·2DMF (M<sup>II</sup>=Mn, Co, Ni) (abbreviated as Fe<sub>2</sub>Mn<sub>2</sub>, Fe<sub>2</sub>Co<sub>2</sub>, and Fe<sub>2</sub>Ni<sub>2</sub>) that have been recently synthesized, within a generalized-gradient approximation in spin-polarized densityfunctional theory (DFT). Due to strong ligand fields present in the molecules, the Fe ions exhibit a low ground-state spin of S = 1/2 in which the orbital angular momentum may not be quenched even without spin-orbit coupling. Based on the calculated electronic structures, the magnetic anisotropy for  $Fe_2Mn_2$ ,  $Fe_2Co_2$ , and Fe<sub>2</sub>Ni<sub>2</sub> is computed including single-electron spin-orbit coupling within a DFT formalism. The theoretical values of the induced orbital angular momentum and of the magnetic anisotropy parameters are compared to those for a single-molecule magnet  $Mn_{12}$ . The total magnetic anisotropy present in the three single-molecule magnets is due to competition between the magnetic anisotropy of the Fe and of the M ions.

<sup>1</sup>This work was supported by the NSF and the Kentucky Science and Engineering Foundation. The computational resources were provided by the TCF at VT and the NCSA.

Kyungwha Park Virginia Tech

Date submitted: 29 Nov 2006

Electronic form version 1.4