

Abstract Submitted
for the MAR07 Meeting of
The American Physical Society

Effects of -NO₂ substitution on charge addition and reorganization energies in phenylene ethynylene oligomers STEVEN ROBEY, NIST-Gaithersburg, N. E. GRUHN, University of Arizona, J. CIZEK, Rice University, J. M. TOUR, Roice University — Reports of non-linear transport in molecular-scale junctions have stimulated suggestions for computing and switching applications based on molecular electronics. One of the most widely referenced results is reported negative differential resistance (NDR) behavior in -NO₂ substituted oligo-phenylene ethynylenes (OPE). Theoretical work has invoked the importance of charge addition effects on conformation and electronic structure and polaronic effects to provide potential explanations for this behavior. We have investigated charge addition effects for pristine versus -NO₂ substituted OPE self-assembled monolayers using photoelectron spectroscopy in combination with “doping” with K. Results are consistent with differences arising from filling of levels associated with the -NO₂ group. We have investigated polaronic effects using photoelectron spectroscopy and optical absorption to guide calculations of reorganization energies for pristine versus -NO₂ substituted OPE’s. We find theoretical evidence for increased reorganization energy with -NO₂ substitution for anionic species by about 33 percent, with experimental values of the reorganization energy ranging from about 0.2 eV to 0.4 eV and theoretical values about 0.2 eV.

Steven Robey
NIST-Gaithersburg

Date submitted: 01 Dec 2006

Electronic form version 1.4