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Hydrogen Adsorption on Nanoporous Biocarbon¹ M.B. WOOD, J.W. BURRESS, C.M. LAPILLI, P. PFEIFER, Dept. of Physics, University of Missouri, Columbia, MO 65211, P.S. SHAH, G.J. SUPPES, Dept. of Chemical Engineering, University of Missouri, Columbia, MO 65211, A.C. DILLON, P.A. PAR-ILLA, National Renewable Energy Laboratory, Golden, CO 80401 — As a part of the Alliance for Collaborative Research in Alternative Fuel Technology (http://allcraft.missouri.edu) we study activated carbons made from corncob, optimized for storing methane and hydrogen (H2) by physisorption at low pressure. We report here: (a) storage capacities of 73-91 g H2/kg carbon at 77 K and 47 bar, validated in three different laboratories (the 2010 DOE target is 60 g H2/kg system); (b) binding energies from H2 adsorption isotherms (c) temperature-programmed desorption data; (d) degree of graphitization of the carbon surface from Raman spectra; (e) pore structure of carbon from nitrogen and methane adsorption isotherms, and small-angle x-ray scattering. The structural analysis shows that the carbon is highly microporous and that the pore space is highly correlated (micropores do not scatter independently).

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