

Abstract Submitted
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Environment of TiO₂ nanoparticles as an important factor to achieve highly efficiency on dye sensitized solar cells¹ TEREZA PARONYAN, M.C. LIN, DEPT. OF APPLIED CHEMISTRY, NATIONAL CHIAO TUNG UNIV. TEAM, INST. FOR PHYSICAL RESEARCH, NAAS, ASHTARAK, ARMENIA COLLABORATION, DEPT. OF CHEMISTRY, EMORY UNIV. COLLABORATION — Amorphous TiO₂ nanoparticles were synthesized by sol gel technique. Environment of nanoparticles was neutralized by ammonia, and pH5.8 of TiO₂ gel was achieved in result, which is close to the point of zero charge (PZC) of anatase TiO₂. Highly interconnected, mesoporous, transparent films were fabricated from the TiO₂ colloid with pH5.8. AFM, SEM XRD analyses were carried out for the investigation the size of nanoparticles, the surface morphology and the crystal structure of films. Volt-amperic characteristics showed an improvement in the cell efficiency along with the increasing of pH TiO₂ colloid. The cells parameters (Voc, FF, Jsc, η) were studied depending on the pH of TiO₂ colloid. Increasing pH of the colloid from 2.1 to 5.8 enhanced the overall conversion efficiency of the dye-sensitized solar cells by approximately 30% , and 9.2 % of efficiency was achieved with N719 dye under illumination by simulated AM1.5 solar light (100 mW/cm²).

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