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Computational model for the production of monodisperse silver spheres in solution¹ DANIEL ROBB, IONEL HALACIUGA, VLADIMIR PRIV-MAN, DAN GOIA, Clarkson University — We report the results of computational modeling of the production of monodisperse, spherical silver particles through the rapid mixing of reducing agent and silver-amine complex solutions, in the absence of a protective colloid. We find that the process can be modeled effectively by a two-stage reaction mechanism used previously to model the production of gold particles [1]. Here, we treat both the equilibrium concentration of silver atoms and the surface tension of silver particles as free parameters in our simulation, finding that the reaction time scale is fit by a narrow region of this parameter space. As in the previous work on gold particles, a kinetic 'dimer suppression factor' is required to limit the number of final particles produced. In addition, we consider an extension of the two-stage reaction model which incorporates the effect of the silver-amine complexation reaction on the availability of Ag monomers. [1] J. Park, V. Privman, and E. Matijevic, J. Phys. Chem. B **105**, 11630 (2001).

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