

Abstract Submitted  
for the MAR09 Meeting of  
The American Physical Society

**Quasielastic Neutron Scattering of Hydrogen Adsorbed in  $\text{KC}_{24}$** <sup>1</sup>

JUSTIN PUREWAL, JAMES KEITH, CHANNING AHN, BRENT FULTZ, California Institute of Technology, CRAIG BROWN, NIST Center for Neutron Research — Quasielastic neutron scattering (QENS) and volumetric techniques were used to study the adsorption of  $\text{H}_2$  by the stage-2 potassium graphite intercalation compound  $\text{KC}_{24}$ . A zero-coverage sorption enthalpy of 8.5 kJ/mol was measured from  $\text{H}_2$  isotherms recorded at 77 K and 87 K. The saturation  $\text{H}_2$  adsorption amount at 77 K was 1.2 mass%, corresponding to a stoichiometry of  $\text{KC}_{24}(\text{H}_2)_{2.0}$ . Quasielastic neutron scattering spectra for  $\text{KC}_{24}(\text{H}_2)_{1.0}$  were collected at temperatures between 40 K and 80 K on a chopper spectrometer and a backscattering spectrometer. Two distinct  $\text{H}_2$  diffusion processes were identified with characteristic times of approximately  $\tau = 10$  ps and  $\tau = 350$  ps at 60 K, respectively. By operating the backscattering spectrometer in fixed window mode, the total elastic scattering of  $\text{KC}_{24}(\text{H}_2)_{1.0}$  was measured as a function of temperature. A sharp decrease in elastic intensity was observed at 35 K due to the onset of quasielastic scattering. This was interpreted as a melting transition of the  $\text{H}_2$  adsorbate in  $\text{KC}_{24}$ .

<sup>1</sup>This work was supported by the Office of Energy Efficiency and Renewable Energy through the Hydrogen Sorption Center of Excellence.

Justin Purewal  
California Institute of Technology

Date submitted: 20 Nov 2008

Electronic form version 1.4