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Rotationally Resolved $\tilde{A}^2 \mathbf{A}_1$ - $\tilde{X}^2 \mathbf{E}$ Electronic Spectra of Symmetric Methoxy Radicals: CH₃O and CD₃O MING-WEI CHEN, DMITRY MEL-NIK, JINJUN LIU, TERRY A. MILLER, The Ohio State University — Methoxy radical has attracted spectroscopic interest for more than twenty years. Microwave measurements of CH₃O and CD₃O with precision on the kHz scale have determined the X^2 E parameters. Jet-cooled laser induced fluorescence (LIF) spectra have also been observed by our group with high-resolution ($\Delta \nu \sim 250$ MHz) and high-accuracy $(\Delta \sigma \sim 50 \text{ MHz})$, for the 3_0^2 and 6_0^1 bands of the $\tilde{A}^2 A_1 - \tilde{X}^2 E_{3/2}$ electronic transition. Since the ground state component $E_{1/2}$ is $\sim 60 \text{ cm}^{-1}$ energetically higher than the $E_{3/2}$ spin component, the $\tilde{X}^2 E_{1/2}$ state is not thermally populated in a jet-cooled environment. However, our complementary stimulated emission pumping (SEP) experiment with the same resolution and accuracy as the LIF work directly interrogates the $\tilde{X}^2 E_{1/2}$ level of CH₃O and CD₃O by depleting the fluorescence from the $\tilde{A}^2 A_1 3^2$ levels. The global analysis of the microwave, LIF, and SEP data breaks correlations in the microwave data and provides better determinations for the \hat{X} and \hat{A} states' parameters. Comparison of the values for CH₃O, ¹³CH₃O, and CD₃O allows us to separate first-order from second-order electronic and vibrational contributions based upon the isotopic dependencies of the effective ground state parameters.

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