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### **Hollandites: a novel class of oxides with unusual properties**

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Hollandite with a general chemical formula  $A_xM_8O_{16}$  ( $x \leq 2$ ) is one of minerals. Its crystal structure consists of the  $M_8O_{16}$ -framework and  $A$ -cations at tunnel sites in the  $M_8O_{16}$ -framework. The most popular crystal symmetry is a tetragonal  $I4/m$  in which the crystallographically independent  $M$  site is unique, hence,  $A_2^+M_8O_{16}$  is a mixed valent oxide with  $M^{3+}/M^{4+} = 1/3$ . We have synthesized  $K_2V_8O_{16}$  [1] and  $K_2Cr_8O_{16}$  [2] in both powder and single crystal form under high pressure and found metal-insulator (MI) transitions. The manner of MI transitions, however, is very different between both compounds.  $K_2V_8O_{16}$  exhibits a first order MI transition at around 170 K. The magnetic susceptibility is reduced to a small value at the transition, suggesting the formation of  $V^{4+}$ - $V^{4+}$  singlet pairs and  $V^{3+}$ - $V^{3+}$  pairs in the low temperature insulator phase. The transition is accompanied by the structural change from a tetragonal to a monoclinic structure. The low temperature phase has a superlattice of  $\sqrt{2} \times \sqrt{2} \times 2$ , suggesting a charge ordering of  $V^{4+}$  and  $V^{3+}$ . On the other hand,  $K_2Cr_8O_{16}$  is a ferromagnetic metal (or half-metal) with  $T_C = 180$  K and shows a transition to an insulator at 95 K without any apparent structural change but retaining ferromagnetism.  $K_2Cr_8O_{16}$  is quite unique in three aspects: It has a rare mixed valence of  $Cr^{3+}$  and  $Cr^{4+}$ ; it has a metal (or half-metal) to insulator transition in a ferromagnetic state; and the resulting low temperature phase is a rare case of a ferromagnetic insulator. This discovery could open a new frontier on the relation of magnetism and conducting properties in strongly correlated electron systems.

References:

- [1] M. Isobe, S. Koishi, N. Kouno, J. Yamaura, T. Yamauchi, H. Ueda, H. Gotou, T. Yagi, and Y. Ueda, J. Phys. Soc. Jpn. **75** (2006) 073801.
- [2] K. Hasegawa, M. Isobe, T. Yamauchi, H. Ueda, J-I. Yamaura, H. Gotou, T. Yagi, H. Sato, and Y. Ueda, Phys. Rev. Lett. **103** (2009) 146403.