

Abstract Submitted
for the MAR10 Meeting of
The American Physical Society

First-Principles Modeling of ThO₂ Solid Solutions with Oxides of Trivalent Cations VITALY ALEXANDROV, Department of Chemical Engineering and Materials Science, and NEAT Organized Research Unit, University of California, Davis, MARK ASTA, Department of Chemical Engineering and Materials Science, University of California, Davis, NIELS GRONBECH-JENSEN, Department of Applied Science, University of California, Davis — Solid solutions formed by doping ThO₂ with oxides of trivalent cations, such as Y₂O₃ and La₂O₃, are suitable for solid electrolyte applications, similar to doped zirconia and ceria. ThO₂ has also been gaining much attention as an alternative to UO₂ in nuclear energy applications, the aforementioned trivalent cations being important fission products. In both cases the mixing energetics and short-range ordering/clustering are key to understanding structural and transport properties. Using first-principles atomistic calculations, we address intra- and intersublattice interactions for both cation and anion sublattices in ThO₂-based fluorite-type solid solutions and compare the results with similar modeling studies for related trivalent-doped zirconia systems.

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Date submitted: 27 Nov 2009

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