## Abstract Submitted for the MAR11 Meeting of The American Physical Society

Origin of the Diverse Melting Behaviors of Aluminum Nanoclusters with Around 55 Atoms<sup>1</sup> JOONGOO KANG, SU-HUAI WEI, National Renewable Energy Laboratory, YONG-HYUN KIM, Korea Advanced Institute of Science and Technology — Microscopic understanding of thermal behaviors of metal nanoparticles is important for nanoscale catalysis and thermal energy storage applications. Using first-principles molecular dynamics simulations, we reveal the microscopic origin of the diverse melting behaviors of  $Al_N$  clusters with N around 55 [1,2]. The conceptual link between the degree of symmetry (e.g.,  $T_d$ ,  $D_{2d}$  and  $C_s$ ) and solidity of atomic clusters is quantitatively demonstrated through the analysis of the configuration entropy. The size-dependent, diverse melting behaviors of Al clusters originate from the reduced symmetry ( $T_d \rightarrow D_{2d} \rightarrow C_s$ ) with increasing the cluster size. In particular, the sudden drop of the melting temperature and appearance of the dip at N = 56 are due to the  $T_d$ -to- $D_{2d}$  symmetry change, triggered by the surface saturation of the tetrahedral  $Al_{55}$  with the  $T_d$  symmetry.

[1] G. A. Breaux, C. M. Neal, B. Cao, and M. F. Jarrold, Phys. Rev. Lett. **94**, 173401 (2005).

[2] J. Kang, S.-H. Wei, and Y.-H. Kim, J. Am. Chem. Soc. (in press).

<sup>1</sup>This work was funded by the U.S. DOE EERE CSP and NREL LDRD programs.

Joongoo Kang National Renewable Energy Laboratory

Date submitted: 21 Nov 2010

Electronic form version 1.4