

Abstract Submitted  
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**Enhancement of Ag Cluster Mobility on Ag(111) Surface by clustering with Chlorine**<sup>1</sup> YU-NING WU, HAI-PING CHENG, Department of Physics and Quantum Theory Project, University of Florida, USA, NOUARI KEBAILI, ALBERT MASSON, CATHERINE BRÉCHIGNAC, Laboratoire Aimé Cotton, CNRS-Université Paris Sud 11, France — Chlorine is observed to accelerate the fragmentation of Ag nanostructures deposited on graphite. To understand the role of chlorine in the stability of Ag nanostructures, we have studied the formation and diffusion of  $\text{Ag}_n$  and  $\text{Ag}_n\text{Cl}_m$  ( $n= 1$  to 4) clusters on Ag(111) surface, using density functional theory (DFT) with generalized gradient approximations (GGA) and the projector-augmented wave method. Our calculation shows that the formation energies and diffusion barriers of  $\text{Ag}_n$  clusters are both lowered by clustering with chlorine when  $n=1, 3$  and 4, indicating the enhancement of mass transport on Ag(111) surface.  $(\text{AgCl})_n$  clusters ( $n=1, 3$  and 4) are good candidates for surface mass transport units. We have also studied a chloridized  $\text{Ag}_{55}$  cluster and an  $\text{Ag}_{55}$ - $\text{Ag}_{55}$  neck structure. Chlorine is found to loosen the  $\text{Ag}_{55}$  structure and weaken the binding between the  $\text{Ag}_{55}$  bulbs in the neck structure.

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