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Preferential condensation of  $\beta$  RDX on In metal surfaces TER-RENCE JACH, NIST, Gaithersburg, MD, ILANA G. GOLDBERG, Transportation Security Laboratory, Atlantic City, NJ, FERNANDO D. VILA, Dept. of Physics, U. of Washington, Seattle, WA — The energetic compound cyclotrimethylenetrinitramine (RDX) normally crystallizes out of solution at standard temperature and pressure in the  $\alpha$  form. This consists of two nitro groups in pseudoaxial positions in relation to the C-N ring, and one nitro group in a pseudoequatorial position in an orthorhombic lattice. A metastable phase, labeled the  $\beta$  phase, is difficult to create and rarely observed. It consists of all three nitro groups in pseudoaxial positions, occupying a trigonal lattice. We have observed by means of Raman spectroscopy that RDX crystallized from solution on In metal foil preferentially adopts the  $\beta$  phase. We discuss a possible mechanism for this behavior in the context of recently published DFT calculations for RDX on a metal cluster.

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