

Abstract Submitted
for the MAR13 Meeting of
The American Physical Society

Catalytic Role of Au Nanowires¹ EDISON DA SILVA, ANA PAULA F. NASCIMENTO, Institute of Physics “Gleb Wataghin”, UNICAMP, 13083-970, Campinas - SP, Brazil, MIGUEL A. SAN-MIGUEL, Physical Chemistry Department, University of Seville, E41012, Seville, Spain — The oxidation of CO in linear atomic chains (LACs) of Au nanowires (NW) is studied by means of density functional theory calculations using quasi-static ($T=0$) and finite temperature *ab initio* molecular dynamics simulations. The adsorption of O₂ and CO molecules on the LAC lead to the formation of an intermediate O₂CO complex. Upon thermal activation at room temperature, the complex is able to proceed to oxidation forming a CO₂ molecule and leaving an atomic O impurity into the Au LAC. We report the conditions under which this oxidation pathway takes place. This process also explains the appearance of unusual large Au-Au bond distances in the LAC and attributed to the presence of atomic impurities.

¹This work is supported by CNPq, CAPES and FAPESP and FAEPEX. APFN was supported by CNPq. CENAPAD-SP and IFGW are acknowledged for computer time.

Edison Z. da Silva
Institute of Physics “Gleb Wataghin”, UNICAMP,
13083-970, Campinas - SP, Brazil

Date submitted: 26 Nov 2012

Electronic form version 1.4