

Abstract Submitted  
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**Defect engineering of complex semiconductor alloys:**  
 $\text{Cu}_{2-2x}\text{M}_x\text{O}_{1-y}\text{X}_y$ <sup>1</sup> STEPHAN LANY, National Renewable Energy Laboratory,  
VLADAN STEVANOVIC, Colorado School of Mines — The electrical properties of  
semiconductors are generally controlled via doping, i.e., the incorporation of dilute  
concentrations of aliovalent impurity atoms, whereas the band structure properties  
(gap, effective masses, optical properties) are manipulated by alloying, i.e., the in-  
corporation of much larger amounts of isovalent elements. Theoretical approaches  
usually address either doping or alloying, but rarely both problems at the same time.  
By combining defect supercell calculations, GW quasi-particle energy calculation,  
and thermodynamic modeling, we study the range of electrical and band structure  
properties accessible by alloying aliovalent cations ( $M = \text{Mg}, \text{Zn}, \text{Cd}$ ) and isova-  
lent anions ( $X = \text{S}, \text{Se}$ ) in  $\text{Cu}_2\text{O}$ . In order to extend dilute defect models to higher  
concentrations, we take into account the association/dissociation of defect pairs and  
complexes, as well as the composition dependence of the band gap and the band  
edge energies. Considering a composition window for the  $\text{Cu}_{2-2x}\text{M}_x\text{O}_{1-y}\text{X}_y$  alloys  
of  $0 \leq (x,y) \leq 0.2$ , we predict a wide range of possible band gaps from 1.7 to 2.6 eV,  
and net doping concentrations between  $p = 10^{19} \text{ cm}^{-3}$  and  $n = 10^{17} \text{ cm}^{-3}$ , notably  
achieving type conversion from p- to n-type at Zn or Cd compositions around  $x =$   
0.1.

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