## Abstract Submitted for the MAR14 Meeting of The American Physical Society

On the Cubic Phase Stability and Magnetic Properties of Cudoped ZrO<sub>2</sub> S. THOTA, Physics Dept., IIT Guwahati, India, K. PISANE, S. SINGH, M.S. SEEHRA, Physics Dept., West Virginia University — Magnetic properties and room temperature cubic-phase stability of Cu-doped  $ZrO_2$  nanocrystallites (16 nm size) with various compositions of  $\operatorname{Zr}_{1-x}\operatorname{Cu}_x\operatorname{O}_{2-x}(0.01 \le x \le 0.25)$ are reported. The samples were synthesized at a constant pH = 8 using zirconyl nitrate hydrate and copper acetate monohydrate as precursors and ethanol as a solvent. Thermal analysis shows that the cubic phase is not stable beyond calcination temperature of 500°C for 8 hours in air and a critical composition  $x_c \approx 0.10$ . For x  $>x_c$ , monoclinic ZrO<sub>2</sub> and CuO emerge as secondary phases with a shrinking unitcell volume for increasing Cu content. Against expectations [1], the temperature and magnetic field dependence of magnetization exhibit no signatures of ferromagnetism down to 2 K. Instead, temperature dependence of magnetic susceptibility for all compositions yields Curie-law variation with a magnetic moment  $\mu \approx 1.3 \mu_B$ per Cu<sup>2+</sup>, which is somewhat smaller than the expected value of  $\mu \approx 1.9 \mu_B$  per  $Cu^{2+}$ . Electron magnetic resonance studies show a signal near  $g \approx 2.1$  due to  $Cu^{2+}$ substituting at  $Zr^{4+}$  sites with four-line hyperfine splitting from the nuclear spin I = 3/2 of Cu.

[1]. S. Ostanin et al, Phys. Rev. Lett. 98, 016101,(2007)

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