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A problem with the stress theorem commonly used in DFT codes¹ DONALD NICHOLSON, Department of Physics, University of North Carolina Asheville — The change in energy when an affine transformation (strain) is applied to a lattice of ions can be obtained by evaluating the algebraic derivative of the DFT energy (in practice a local or other approximation) of an electron density that has been similarly strained [1]. Because the DFT energy is stationary in the density, it is only required that the strained density reduces to the exact density at zero strain; it does. The algebraic derivatives of the Hartree and exchange energies are straightforward. The derivative with respect to strain of the non-interacting kinetic energy depends on two assumptions: 1) the modulus squared of the strained orbitals equals the strained electron density, and 2) the strained orbitals minimize the non-interacting kinetic energy. The first assumption is correct. I find that the second assumption applies only in special cases. The limitations and possible modifications of the stress theorem are discussed.

[1] Nielsen, O. H. & Martin, R. M. 1983 First-Principles Calculation of Stress. Physical Review Letters 50, 697.

[2] D. M. Nicholson, Madhusudan Ojha, and T. Egami, Journal of Physics Condensed Matter 10/2013; 25(43):435505.

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