

Abstract Submitted
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**Bond pathway analysis of NMR spectra for $\text{Li}_{1.2}\text{Mn}_{0.4}\text{Co}_{0.4}\text{O}_2$:
pristine material** HAKIM IDDIR, BARIS KEY, FULYA DOGAN, JOHN RUSSELL, BRANDON LONG, JAVIER BARENO, JASON CROY, ROY BENEDEK, Argonne National Laboratory — NMR has been applied extensively to lithium ion battery cathode materials, of which layered-layered composites $x\text{Li}_2\text{MnO}_3 \cdot (1-x)\text{LiMO}_2$ ($M = \text{Mn}, \text{Co}, \text{Ni}$) are of particular interest, owing to their high energy density. In this work, NMR spectra are measured for the model layered-layered system $x\text{Li}_2\text{MnO}_3 \cdot (1-x)\text{LiCoO}_2$ and Bond-Pathway-model analysis is applied to elucidate the atomic arrangement and domain structure of this material (in its pristine state, before electrochemical cycling). The simplest structural element of a domain consists of a stripe of composition LiMn_2 parallel to an in-layer crystallographic axis in a metal layer of the composite. A simple model of the composite structure may be constructed by a superposition of such stripes in an LiCoO background. We show that such a model can account for most of the features of the observed NMR spectra. Support from the Vehicle Technologies Program U.S. Department of Energy, Office of Energy Efficiency and Renewable Energy.

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