MAR16-2015-000247

Abstract for an Invited Paper for the MAR16 Meeting of the American Physical Society

Electrons and phonons in layered and monolayer vanadium pentoxide.¹ WALTER R. L. LAMBRECHT, Case Western Reserve University

Vanadium pentoxide (V₂O₅) is a layered material with the potential for interesting new properties when made in 2D monoor few-layer form. Its band structure is characterized by a split-off conduction band. The lowest conduction band is separated from the rest of the conduction bands by about 1 eV and consists of V- d_{xy} orbitals, non-bonding to the oxygens by symmetry. This narrow band has dispersion essentially along the direction of chains occurring in the layer. When this band becomes half-filled by doping, spin-splitting occurs accompanied by an antiferromagnetic coupling between nearest neighbors along the chain direction. This situation is well known to occur in the so-called ladder compound NaV₂O₅, which was extensively studied in the late 90s as a potential spin-Peierls or charge ordering compound. However, the monolayer form of V₂O₅ may allow for other ways to control the doping by gating, removing vanadyl oxygens, adsorption of alkali metals, nanoribbon formation, etc. Our calculations predict a switch from antiferromagnetic to ferromagnetic coupling for doping slightly less than half filling of the split-off band. In this talk we will discuss our recent work on the electronic band structure of both bulk and monolayer V₂O₅ as well as the phonons. We find that the quasi-particle self-consistent *GW* method strongly overestimates the band gap. Lattice polarization corrections of the screening are required because of the large LO/TO phonon frequency ratios. Excitonic effects may also be expected to be fairly large. We find that some of the vibrational modes, notably the vanadyl-oxygen bond stretch perpendicular to the layer, unexpectedly shows a strong blue shift. This is explained in terms of reduced screening affecting the long-range dipole components of the force constants.

¹Supported by AFOSR and DOE. Work done with Churna Bhandari, Mark van Schilfgaarde and Andre Schleiffe.