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First Principles Investigation of the Geometrical and Electrochemical Properties of $\text{Na}_4\text{P}_2\text{S}_6$ and $\text{Li}_4\text{P}_2\text{S}_6$.¹ LARRY E. RUSH JR., N.A.W. HOLZWARth, Wake Forest University — First principles simulations are used to examine the structural and physical properties of $\text{Na}_4\text{P}_2\text{S}_6$ in comparison with its $\text{Li}_4\text{P}_2\text{S}_6$ analog. Four model structures are considered including the $C2/m$ structure recently reported by Kuhn and co-workers² from their analysis of single crystals of $\text{Na}_4\text{P}_2\text{S}_6$, and three structures related to the $P6_3/mcm$ structure with P site disorder found in 1982 by Mercier and co-workers³ from their analysis of single crystals of $\text{Li}_4\text{P}_2\text{S}_6$. The computational results indicate that both $\text{Na}_4\text{P}_2\text{S}_6$ and $\text{Li}_4\text{P}_2\text{S}_6$ have the same disordered ground state structures consistent with the $P6_3/mcm$ space group, while the optimized $C2/m$ structures have higher energies by 0.1 eV and 0.4 eV per formula unit for $\text{Na}_4\text{P}_2\text{S}_6$ and $\text{Li}_4\text{P}_2\text{S}_6$, respectively. Simulations of ion migration suggest that $\text{Na}_4\text{P}_2\text{S}_6$ may have more favorable ionic conductivity compared to $\text{Li}_4\text{P}_2\text{S}_6$.

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²ZAAC **640**(5):689-692 (2014)

³JSSC **43**:151-162 (1982)

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