

MAR17-2016-009809

Abstract for an Invited Paper
for the MAR17 Meeting of
the American Physical Society

Multi-Scale Structure of Coacervates formed by Oppositely Charged Polyelectrolytes¹

MICHAEL RUBINSTEIN, University of North Carolina at Chapel Hill

We develop a scaling model of coacervates formed by oppositely charged polyelectrolytes and demonstrate that they self-organize into multi-scale structures. The intramolecular electrostatic interactions in dilute polyanion or polycation solutions are characterized by the electrostatic blobs with size D_- and D_+ respectively, that repel neighboring blobs on the same chains with electrostatic energy on the order of thermal energy kT . After mixing, electrostatic intramolecular repulsion of polyelectrolytes with higher charged density, say polyanions, keeps these polyanions in coacervates aligned into stretched arrays of electrostatic blobs of size $D_- < D_+$ on length scales smaller than inter-polyanion distance $\xi_- \approx D_-^{1/4} D_+^{3/4}$ in a θ -solvent for uncharged backbones, while their conformations are random walks on larger length scales. Weaker charged polycations adsorb on stronger charged polyanions forming a screening “coat” around them analogous in its structure to a semidilute solution of uncharged polymers with correlation length $\xi_+ \approx D_-^{-1/8} D_+^{9/8}$ and random walk conformations on all length scales. In this coat, the electrostatic attraction of polycations to polyanions is balanced by the short-range repulsion between sections of polycations. This attraction slowly decreases in strength with increasing distance from a polyanion due to screening by the polycation “coat” resulting in a slow decrease of coat concentration. The inter-polyanion correlation length ξ_- is determined by the length scale at which the polycation “coat” compensates the polyanion charge. The relative values of length scales in the coacervate are $D_- < \xi_+ < \xi_-$. The opposite charges form bound pairs and larger ionic clusters at higher strength of electrostatic interactions resulting in the formation of mixed ionic/”screening coat” coacervates and eventually at the highest electrostatic strength – in ionic networks of comb/bottlebrush-like polyanion backbones with polycation loops and bridges. These coacervate networks are super-tough, can be used for microencapsulation and drug delivery, and can also function as underwater glue.

¹This work was done in collaboration with Drs. Sergey Panyukov and Qi Liao and supported by NSF, NIH and CFF.