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Infrared Studies of Titanium-Containing Vitreous Silica¹ CAR-RIE E. BLACK, University of Vermont, CHRISTOPHER C. HOMES, Brookhaven National Laboratory, DAVID Y. SMITH, University of Vermont and Argonne National Laboratory — The reflectivity of samples of Corning's ULE glass (7.5% Ti)and 92.5% silica), was measured from 18 cm⁻¹ to 30000 cm⁻¹. The refractive index and extinction coefficient were then calculated using the Jahoda-Velicky dispersion relation. The resulting infrared absorption spectrum is almost identical to that of vitreous silica with the exception of a new absorption band at 941 $\rm cm^{-1}$. The negligible change in the vitreous silica part of the absorption is taken to imply that, at the concentrations present, titanium has little effect on the SiO_2 network and the environment of Si⁴⁺ ions. The phase diagram for TiO₂-SiO₂ systems and titanium EXAFS measurements² indicate that titanium is tetrahedrally coordinated by O^{2-} ions in ULE glass, i.e., it substitutes for Si^{4+} . On this basis the new 941 cm⁻¹ band is attributed to an antisymmetric stretching mode of Ti-O-Si units, corresponding to a perturbation of the prominent Si-O-Si antisymmetric stretching mode³ seen at 1076 cm^{-1} in vitreous silica. 2. R. B. Greegor, *et al.* J. Non-Crystalline Solids 55, 27 (1983). 3. C. T. Kirk, Phys. Rev. 38, 1255 (1988).

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