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Classical Trajectory Studies of the Hydrogen Peroxyl Radical
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Columbia — The intramolecular dynamics, intramolecular vibrational redistribu-
tion of energy (IVR), isomerization and unimolecular dissociation of the hydrogen
peroxyl radical, $\text{HO}_2^* \rightarrow \text{H} + \text{O}_2$ have been studied using classical trajectories.
Mode specific effects are found to affect the IVR, isomerization and the rate of dis-
sociation. Exchange of the hydrogen atom between the oxygen atoms increases the
rate of energy transfer between the vibrational modes of the radical. The relaxation
mechanism of the vibrationally excited radical embedded in a dense gas environment
is also presented.

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