

Abstract Submitted  
for the SHOCK05 Meeting of  
The American Physical Society

**The Stability of  $\text{MgSiO}_3$  Perovskite at Lower Mantle Conditions<sup>1</sup>**

LI ZHANG<sup>2</sup>, ZIZHENG GONG, XIUFANG CHEN, LIWEI DONG, Institute of Physics, Southwest Jiaotong University, Chengdu 610031, P.R. China, YINGWEI FEI, Geophysical Laboratory, Carnegie Institution of Washington, Washington DC 20015, USA, FUQIAN JING, Laboratory for Shock Wave and Detonation Physics Research, Institute of Fluid Physics, P.O.Box 919, Mianyang, Sichuan 621900, China — The stability of  $\text{MgSiO}_3$ -perovskite at lower mantle conditions has been a subject of extensive investigation and debate. Shock recovery experiments with the initial sample of  $(\text{Mg}_{0.92}, \text{Fe}_{0.08})\text{SiO}_3$  enstatite and  $\text{MgO}+\text{SiO}_2$  were conducted up to 120GPa. The analysis of XRD results indicate that there is no possibility for the chemical decomposition from  $(\text{Mg}_{0.92}, \text{Fe}_{0.08})\text{SiO}_3$  to oxides  $\text{SiO}_2$  and  $(\text{Mg}_{0.92}, \text{Fe}_{0.08})\text{O}$  under experimental shock pressure. The Gibbs energy and molar volume of all phases in the reaction  $\text{MgSiO}_3(\text{Pv}) = \text{MgO}(\text{Pe}) + \text{SiO}_2(\text{St})$  are calculated using both the latest experimental thermodynamic parameters and the first principle molecular dynamic (MD) simulations under lower mantle conditions (1000-3500K and 30-150GPa)., and the results show that perovskite is thermodynamically stable relative to the stishovite and periclase assemblage at lower mantle conditions.

<sup>1</sup>This research was supported by the National Natural Science Foundation of China under Grant No. 10299040 and 40474033.

<sup>2</sup>Now at: Geophysical Laboratory, Carnegie Institution of Washington, Washington DC 20015, USA.

Zizheng Gong  
Institute of Physics, Southwest Jiaotong University  
Chengdu 610031, P.R. China.

Date submitted: 16 Mar 2005

Electronic form version 1.4