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Equations-of-State and Shock Response of Fluorinated Polymers: Influence of Polymer Crystallinity DANA DATTELBAUM, DAVID ROBBINS, STEPHEN SHEFFIELD, EDWARD ORLER, RICK ALCON, DAVID STAHL, LANL, DETONATION PHYSICS TEAM — Fluoropolymers find wide application in a variety of fields due to their chemical inertness, low coefficient of friction, and ability to withstand high-temperature operating conditions. The presence of fluorine in the polymer backbone, coupled with the propensity of these polymers to adopt linear chains, often result in the presence of complex, multi-phase crystalline structures. We will examine the influence of the degree of crystallinity, and nature of crystalline phase on both static compressive and shock loaded polymer behavior in poly(tetrafluoroethylene) (PTFE), poly(chlorotrifluoroethylene), and poly(chlorotrifluoroethylene-co-vinylidene fluoride). The most widely used fluorocarbon polymer for engineering applications is poly(tetrafluoroethylene) (PTFE). PTFE exists in at least four known crystalline phases near room temperature and ambient pressure. Replacing one of the fluorines in the chemical backbone with chlorine results in a polymer with substantially different properties, poly(chlorotrifluoroethylene) or Kel-F 81. The third, related polymer that will be discussed is the high explosive binder, poly(chlorotrifluoroethylene-co-vinylidene fluoride) (Kel-F 800) that is used in PBX-9502 formulations. Accompanying static high-pressure diamond anvil cell work will also be presented and compared to dynamic results.

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