

Abstract Submitted
for the TSF12 Meeting of
The American Physical Society

Theoretical Study of the Properties of the Type II Clathrate A^xSn^{136} , ($A = \text{alkali atom}$; $0 \leq x \leq 24$) DONG XUE, CRAIG HIGGINS, CHARLEY MYLES, Texas Tech University — Motivated by recent experimental and theoretical interest in the x dependence of the properties of the Si and Ge-based Type II clathrate materials A^xSi^{136} and A^xGe^{136} ($A = \text{alkali atom}$) [1,2] we are carrying out a systematic theoretical study of the properties of the Sn-based Type II clathrate system A^xSn^{136} . Type II clathrates have cage-like lattices in which Si, Ge, or Sn atoms are tetrahedrally-coordinated and sp_3 covalently bonded. The cages can contain “guests”; usually alkali or alkaline earth atoms. These materials are particularly interesting because of their potential use as thermoelectrics. Recent powder X-ray diffraction experiments have found the very interesting result that, for increasing x in the range $0 \leq x \leq 8$ a lattice contraction occurs, and that x is increased further ($8 \leq x \leq 24$), a contrasting lattice expansion results. These observations have motivated us to study the behavior of the lattice constant and other properties as a function of guest concentration in other Type II clathrates. In the present paper, we report preliminary results of a density functional based theoretical study of the properties of K^xSn^{136} as a function of x . We present results for the x dependence of the lattice constant as well as for other structural and electronic properties of this material.

[1] S. Stefanoski and G. Nolas, Cryst. Growth Des. 2011, [dx.doi.org/10.1021/cg200756r](https://doi.org/10.1021/cg200756r)

[2] M. Beekman, E. Nenghabi, K. Biswas, C. Myles, M. Baitinger, Y. Grin, G.S. Nolas, Inorg. Chem. 49 2010, DOI: 10.1021/ic1005049

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Date submitted: 21 Sep 2012

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