

Abstract Submitted
for the GEC10 Meeting of
The American Physical Society

Synthesis and Electrical Characterization of *n*-Type Nanocrystalline Diamond Films by Microwave Plasma-Enhanced Chemical Vapor Deposition KUNGEN TEII, TOMOHIRO IKEDA, Kyushu University — Nanocrystalline diamond (nanodiamond) films are composed of three different carbon phases: the diamond phase in form of nano-sized grains, trans-polyacetylene segments, and amorphous carbon. They are typically formed under C₂-rich conditions by CVD in Ar-rich/CH₄ plasmas. *n*-type conductivity in nitrogen-incorporated nanodiamond films is attributed to the formation of electronic states associated with carbon and nitrogen in the grain boundary. However, the origin of the high *n*-type conductivity still remains unclear. The authors investigate structure and electrical properties of *n*-type nanodiamond films prepared from a microwave Ar-rich/N₂/CH₄ plasma. The authors also investigate the rectification properties of *p* – *n* diodes using *n*-type nanodiamond films. The plasma was characterized by strong emission from C₂ radicals. The room-temperature resistivity of the films decreased exponentially by three orders of magnitude with deposition temperature and was saturated at $\sim 10^{-2} \Omega \text{ cm}$. The electron concentration increased up to 10^{20} cm^{-3} , while the mobility was between 1 and $10 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$. Arrhenius plots of the conductivity showed a transition from semiconducting to quasi-metallic conduction with deposition temperature. The amount and clustering of the sp² phase were found to affect strongly the electrical conduction properties.

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Date submitted: 12 Jun 2010

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