

Abstract Submitted  
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**First-principles simulations of extended phosphorus oxynitride structures in LiPON glasses**<sup>1</sup> YAOJUN DU, N. A. W. HOLZWARTH, Wake Forest University — The thin film electrolyte LiPON, having the composition of  $\text{Li}_{3+x}\text{PO}_{4-y}\text{N}_z$  with  $x = 3z - 2y$ , was developed at Oak Ridge National Lab in the 1990's for use in solid state batteries and related applications.<sup>2</sup> In an effort to understand and to optimize properties of this electrolyte material, we expanded previous studies of isolated defects in crystalline  $\text{Li}_3\text{PO}_4$ <sup>3</sup> to focus on more complicated phosphate structures based on combinations of tetrahedral P–O bonds and bridging P–O–P bonds. For example, crystalline  $\text{LiPO}_3$ <sup>4</sup> and  $\text{P}_2\text{O}_5$ <sup>5</sup> are composed of phosphate structures with linear and branched chains, respectively. Both these and related structures derived from substituting O with N and adjusting mobile Li ion concentrations approximate components found in LiPON films.<sup>2</sup> In the simulated structures, we find that N is energetically more stable at bridging bond sites than at tetrahedral sites by 2-3 eV and that the Li ion migration energies are 0.5–0.6 eV, similar to values measured in LiPON films.

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<sup>2</sup>N. J. Dudney, *Interface* **17**:3, 44 (2008) and listed references.

<sup>3</sup>Y. A. Du and N. A. W. Holzwarth, *Phys. Rev. B* **78**, 174301 (2008).

<sup>4</sup>E. V. Murashova and N. N. Chudinova, *Cryst. Rept.* **46**, 942 (2001).

<sup>5</sup>E. H. Arbib and co-workers, *J. Solid State Chem.* **127**, 350 (1996).

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