

Abstract Submitted  
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**$^3\text{He}$ - $^4\text{He}$  liquid mixtures investigated by neutron imaging technique at low temperatures**<sup>1</sup> PATRICK GUMANN, University of Waterloo, JULIA SCHERSCHLIGT, DANIEL HUSSEY, DAVID JACOBSON, NIST Center for Neutron Research, DAVID CORY, IVAR TAMINIAU, University of Waterloo — Helium is a unique element which exhibits a variety of different phases and unusual behaviors. It can be found in nature in two stable isotopic forms:  $^3\text{He}$  and  $^4\text{He}$ . One of the most profound quantum mechanical effects, superfluidity, occurs below 2.17 K in liquid helium  $^4\text{He}$  and 0.003 K in liquid  $^3\text{He}$ . There are also interesting phenomena occurring in mixtures of the two isotopes. One demonstrative example is the finite solubility of liquid  $^3\text{He}$  (a Fermi system) in superfluid  $^4\text{He}$  (a Bose system) even at  $T = 0$  K. This is the basic principle in the operation of a  $^3\text{He}$ - $^4\text{He}$  dilution refrigerator capable of continuously producing 2 mK. While much has been done in studies of the thermodynamical, quantum properties of liquid helium mixtures, there has not been any attempt to visualize the dynamics of  $^3\text{He}$  in liquid  $^4\text{He}$ . Presented results of neutron imaging experiments on 0.3 bar liquid  $^3\text{He}$ - $^4\text{He}$  mixtures, at 1.5 K have shown a clear diffusion of  $^3\text{He}$  driven by the difference in chemical potential. The data were taken for over 12 hours using a high resolution CCD camera.

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