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***In Situ* Correlation of Volumetric Expansion with Charge Storage in Nanostructured MnO<sub>2</sub>** TETYANA IGNATOVA, BRAD CORSO, DENG PAN, O. TOLGA GUL, PHILLIP G. COLLINS, Department of Physics and Astronomy, University of California at Irvine — Pseudocapacitors aim to meet developing energy storage needs by combining the high energy density of batteries with the power performance of capacitors. However, degradation remains a critical issue for pseudocapacitor electrodes. After many cycles, nanostructured metal oxides like MnO<sub>2</sub> lose their capacity through mechanisms that remain poorly understood. In this work, we studied the volumetric changes that accompany charge storage in nanoscale MnO<sub>2</sub> electrodes by combining in-liquid atomic force microscopy (AFM) with 3-terminal electrochemical cycling. Typical samples consisted of thin films (100 to 400 nm) of porous, amorphous MnO<sub>2</sub> deposited onto 2  $\mu\text{m}^2$  Pt electrodes and then cycled in aqueous LiClO<sub>4</sub> electrolyte. *In situ* measurements of film expansion during charge insertion observed 3.6% volumetric expansion for partial charging of 0.1 electron per Mn atom over a wide range of scan rates and voltage windows, even though these parameters change the balance between fast, double-layer capacitance and bulk, redox pseudocapacitance mechanisms. In fact, volume expansion is universally attributed to bulk charging, so the invariance highlights an unexpected role for surface processes in nanostructured electrode materials.

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