

Abstract Submitted  
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**Investigation of Local Structure and Cation Ordering in Dielectric Oxide Microwave Ceramics with stoichiometry  $A(\text{Li}_x(\text{Nb},\text{Ta})_y)\text{O}_3$  Using  $^7\text{Li}$  and  $^{93}\text{Nb}$  solid-state NMR spectroscopy.** RONY KALFARISI, NMR Spectroscopy Group, Physics Department, College of William and Mary — The local structure and cation ordering in dielectric oxide microwave ceramics with stoichiometry  $A(\text{Li}_x(\text{Nb},\text{Ta})_y)\text{O}_3$  are investigated using  $^7\text{Li}$  and  $^{93}\text{Nb}$  solid-state NMR spectroscopy. For all samples,  $^7\text{Li}$  MAS NMR spectra show one strong and sharp resonance peak indicating one unique environment which corresponds to local lithium environment of nearest B-site neighbor (nBn) surrounded by 4  $\text{LiO}_6$  octahedra and 2  $\text{NbO}_6$  octahedra ( $\text{TaO}_6$  in some samples). In addition to this,  $^7\text{Li}$  MAS NMR spectrum of  $(\text{Ca}_{2/3}\text{La}_{1/3})(\text{Li}_{1/3}\text{Nb}_{2/3})\text{O}_3$  show one additional weak and broad resonance peak which can be assigned to nBn of 3  $\text{LiO}_6$  octahedra and 3  $\text{NbO}_6$  octahedra.  $^{93}\text{Nb}$  MAS NMR spectra of samples with niobium content, show a resonance peak with tail toward the low frequency limit, an evidence to the existence of chemical shifts and quadrupole couplings distributions. Both  $(\text{Sr}_{2/3}\text{La}_{1/3})(\text{Li}_{1/3}\text{Nb}_{2/3})\text{O}_3$  and  $\text{Ca}(\text{Li}_{1/4}\text{Nb}_{3/4})\text{O}_3$  spectra show one broad resonance peak, which can be interpreted as one  $\text{NbO}_6$  octahedron nBn with many slight variations through out the sample. While  $(\text{Ca}_{2/3}\text{La}_{1/3})(\text{Li}_{1/3}\text{Nb}_{2/3})\text{O}_3$  spectra show four peaks correspond to four distinct  $\text{NbO}_6$  octahedra local nBn environments with the nBn configuration as: (i) 3  $\text{LiO}_6$  and 3  $\text{NbO}_6$ ; (ii) 2  $\text{LiO}_6$  and 4  $\text{NbO}_6$ ; (iii) 1  $\text{LiO}_6$  and 5  $\text{NbO}_6$ ; (iv) all 6  $\text{NbO}_6$

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