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Conversion of CO₂ into Useful Fuels using Cu_x/TiO₂ Photocatalysts¹ SATISH KUMAR IYEMPERUMAL, AARON N. DESKINS, worcester polytechnic institute, NA TEAM — Conversion of carbon dioxide, an abundant greenhouse gas, into useful fuels can help solve issues associated with both energy and the environment. Experiments have successfully shown activity for CO₂ conversion to products like methanol using Cu/TiO₂ photocatalysts. How this catalyst works and how it could be improved is an area of much research. We studied this catalyst using density functional theory (DFT) to obtain atomic level insights in the CO₂ reduction process on the catalyst surface. A key activation step in CO₂ reduction is the formation of CO₂ anion species with a bent structure. We modeled small Cu_x (x=1-4) clusters on a TiO₂-anatase surface. Our results show that Cu is able to activate CO₂ into a bent configuration that can be further reduced. Charge analysis indicates that CO₂ does indeed become negatively charged in a bent configuration, but not in a linear adsorption mode. We analyzed charge on Cu to assign its oxidation state, as well as calculating adsorbed CO vibrational modes, a common experimental method to assign oxidation state of supported metals. Our results identify how Cu clusters on TiO₂ surfaces may activate CO₂. Such knowledge is crucial towards refining and designing better catalysts for CO₂ reduction.

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satish kumar iyemperumal
worcester polytechnic institute

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